

ENERGY AND ENVIRONMENT CARBON CAPTURE & SEQUESTRATION

TG/ma

5.4

-5.8

-e o

Time/min

AH = 218.1 m.I

∆m = +298 µg

120

90

100% He

150

∧m = -236 µg

Adsorption and desorption of CO₂ on a catalyst

INTRODUCTION

CO₂ emissions have become nowadays a major concern. One way to limit CO₂ concentration in atmosphere is to adsorb it on specific material. CALVET PRO TG-DSC with its sensitivity which is not depending on the nature of the

∆H = -233 m J

90% He + 10% CO.

60

Heat Flo wmW

≙

1.5

1.0

100% He

gas is a particularly powerful tool as it makes possible to study adsorption and desorption of a gas on a catalyst.

EXPERIMENT

- Sample : catalyst
- Mass : 40 mg
- Temperature : 40°C

• Atmosphere : pure helium during 5 min then 10% CO₂ in helium during 2 hours, then pure helium again during one hour.

Note : before experiment, the catalyst was previously heated up to 220°C during 2 hours under pure helium.



When the mixture 10% CO₂ + Helium is introduced, an adsorption is observed : it corresponds to a mass increase of 298 µg and an exotherm of 233 mJ. When the gas is pure helium again, a desorption is observed: it corresponds to a mass loss of 236 µg and an endotherm of 218.1 mJ.

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INSTRUMENT



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