

# **ENERGY AND ENVIRONMENT** CARBON CAPTURE & SEQUESTRATION

## CO<sub>2</sub> adsorption into zeolite 13X with GASPRO

### INTRODUCTION

The rising level of CO<sub>2</sub> in atmosphere has been linked to global warming. To mitigate the global warming, R&D is being directed towards understanding the relevant phenomena and foster innovation in the field of CO<sub>2</sub> capture and sequestration (CCS). Due to their well-controlled pore structure and size, zeolites have been primary candidates in the gas separation (e.g. CO<sub>2</sub> capture) in industry. Knowledge about the CO<sub>2</sub> sorption properties of zeolites (adsorption capacity, pressure regimes and kinetics) is essential to the design of advanced materials capable of capturing CO<sub>2</sub> in industrial settings. Among zeolites, 13X is known for its relatively high CO<sub>2</sub> capacity. This application note highlights precision measurements of the absorptive properties of a zeolite 13X over a wide range of temperatures.

#### **EXPERIMENT**

CO<sub>2</sub> adsorption into zeolite 13X was measured at various temperatures using a GASPRO Sievert's apparatus which was developed to study sorption of a variety of gases from vacuum up to 200 bar and from liquid He to 500 °C Temperatures. Gas density temperature correction were done by measuring the apparent free gas volume at temperature.

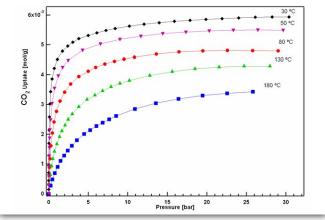


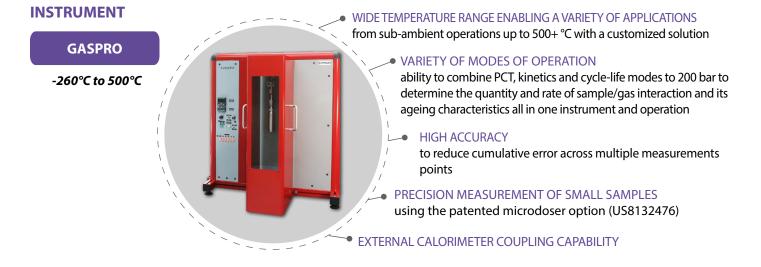
Figure 1. PCT- isotherms for zeolite 13X at 30, 50, 80, 120 and 180 °C

#### **RESULTS AND CONCLUSION**

#### The PCT isotherms for CO2 adsorption into zeolite

13X are shown in Figure 2. The zeolite capacity decreases with temperature reflecting the physisorption nature of the adsorption isotherms. The data are in good agreement with literature. For example, the CO<sub>2</sub> capacity at 30 °C 20 bar is 5.7 moles/kg (5.0-6.4 moles/kg in the literature). The GASPRO is well-suited for the detailed characterization of materials used in CCS (adsorption of CO<sub>2</sub> onto different solid sorbents). The ease of use and the temperature and pressure range are ideal for this type of materials application.

Reference: R Siriwardane, M Shen, E Fisher, et al. NETL report, www.netl.doe.gov



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