

## INORGANIC MATERIALS SCIENCES METALS & ALLOYS

## Cyclic oxidation of metallic alloys by symmetrical TGA

### **INTRODUCTION**

Metallic alloys are frequently subjected to combined environmental attack and mechanical stresses during their actual service life. Their resistance to this complex loading relies partly on their ability to form a protective oxide scale, i.e. an oxide layer with low growth kinetics and high adherence to the alloy. The nature and kinetics of the growth of the oxide layer is, often studied using isothermal laboratory tests. When testing the resistance to oxidation of high-temperature materials, the cyclic-oxidation test is used as a reference because it integrates isothermal oxidation kinetics, oxide-scale adherence, mechanical stresses, metallic alloy and oxide creep and the evolution of these properties with time, for conditions close to the actual conditions of use.

#### **Description**

THEMYS DUO is a symmetrical analyzer with a top loading balance.

This configuration allows to hang the metallic sample (sheet, disk, ribbon...) to the balance without using a crucible in order to increase the contact surface for the gas interaction.

Different hanging systems are available according to the shape of the probe.







#### **EXPERIMENT**

A NiAl single crystal was oxidized in a THEMYS DUO. The experiment consisted of 168 cycles including dwell times of 5 min at 1150°C and 30 min dwell times at 150°C. Heating was at a constant rate of 60°C/min and cooling was controlled at 20°C/min down to 150°C.

The experiment was conduced in flowing oxygen. The obtained Net Mass Gain recording, without any data processing, is given in Fig. 2. This curve includes 12,000 data points. The usual shape of cyclic oxidation data combining mass gain due to oxidation and mass loss due to spalling is obtained.

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#### **RESULTS AND CONCLUSION**

The recorded signal is detailed in Fig. 3 which is an enlargement of Fig. 2 during two cycles (148th and 149th) selected at random for illustration. On this enlargement, it is seen that the mass gain during the high-temperature dwell is about 10 mg at 1150°C.



#### Reference: D. Monceau and D. Poquillon, Oxidation of Metals, Vol. 61, Nos. 1/2, Feb. 2004



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